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#### LOW VOC WEB OFFSET HEATSET INKS

#### FIELD OF THE INVENTION

The invention relates to a novel Web Offset heatset ink composition that dries at higher speeds and contains a latex polymer with amine functionality and having less than 2% by weight of volatile organic compounds.

#### BACKGROUND OF THE INVENTION

Historically, lithographic web offset heatset inks contain between 30 and 45% volatile organic compounds (VOC). As the VOC have detrimental effects on the environment, it is desirable to reduce the VOC content as much as possible. Initial attempts at solving this problem involved the use of chemical reactions that were triggered in the press oven. However, any systems that led to cure did not have shelf stability.

Therefore, typically, a heat setting web offset ink will contain the following major components:

- 1. A high molecular weight ink resin. This material provides the toughness and gloss the ink requires on drying. It also helps to disperse the pigment.
- 2. Solvents: The solvent provides the fluidity to the ink before it is placed on the web and dried in an oven.
  - 3. Pigment.
- 4. Other minor components can include: gallants, which provide structure to the ink, plasticizers (non volatile solvents), waxes, driers, thickeners, antioxidants.

The ink sets or dries by evaporation (and to some degree by penetration of the ink oil into the paper) of the ink oil on heating at 250-300°F, leaving behind a hard polymeric film.

EP 731150 A1 960911 describes rapid thermosetting low VOC web offset lithographic ink systems comprising solid resin, drying oil alkyds, bodied drying oil, vegetable oil, fatty acids, multifunctional unsaturated polyester, reducing agents and transition metal salts of organic acids. Also listed as part of that ink system is an aqueous fountain solution containing peroxides that promote free radical polymerization of the ink. WO96/34922, U.S. Patent 5,431,721, and U.S. Patent 5,545,741, 1996 describe lithographic inks which employ non-volatile solvents and set by penetrat ion of the non-volatile solvent into the stock.

Despite the teachings of the prior art, there is still a need to formulating low VOC web offset heatset inks that has good shelf stability and high dry speed.

## 40 **SUMMARY OF THE INVENTION**

The present invention provides a Web Offset heatset ink composition comprising an aqueous polymer latex dispersed in an ink base that comprises:

(a) an ink resin;

- (b) a non-volatile plasticizer; and
  - (d) a pigment;

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wherein said polymer latex has amine functional groups and said ink had less than about 2 percent by weight of volatile organic compounds (VOC).

The present invention also provides a method for increasing drying or setting speed of a Web Offset heatset ink composition which has less than about 2 percent by weight of volatile organic compounds (VOC) and which comprises:

- (a) an ink resin;
- (b) a non-volatile plasticizer; and
- (d) a pigment;

said method comprising adding to said ink composition an aqueous polymer latex having amine functional groups.

The present invention further provides a method of increasing shelf stability of a Web Offset heatset ink composition which has less than about 2 percent by weight of volatile organic compounds (VOC) and which comprises:

- 20 (a) an ink resin;
  - (b) a non-volatile plasticizer; and
  - (d) a pigment;

said method comprising adding to said ink composition an aqueous polymer latex having amine functional groups and a protective colloid which comprises acid functional groups.

Other objects and advantages of the present invention will become apparent from the following description and appended claims.

### **DETAILED DESCRIPTION OF THE INVENTION**

Low VOC (less than about 2% and preferably about 0% VOC) web offset heatset inks which print cleanly on conventional lithographic plates and dry at press speeds of at least 1000 ft/minute under typical printing conditions are described. The inks consist of a polymer latex dispersed in an ink base made up of ink resins, a non- volatile plasticizer (preferably ethylhexyltallate), and the pigment. The key attribute which leads to shelf stability, but allows for heatsetting on the press, is the use of a water soluble, acid functionalized protective colloid (support resin) which protects the latex particle from interacting with the ink resin and plasti cizer on the shelf, but collapses and allows for interaction in the oven of the press.

The polymer latex employed by the present invention (preferably acrylic:styrene copolymer latex) has amine functionality which results in a faster setting ink system when compared to a non-amine functional latex. The ink minus the latex is a viscous oil, and as it is non volatile, it will not dry (or set) under heatsetting conditions. With the latex present, once the ink reaches the oven, water and ammonia evaporate, and heat forces collapse of the glassy latex particles so that they may blend with the rest of the ink, giving rise to a hard film. The

amine groups in the latex can react with acid groups (forming salts) in the ink resins upon collapse of the latex in the oven and give rise to an even tougher film. Accordingly, this increase in setting speed is believed to be due to the attraction between the amine functional latex and the acid functional ink resins that occurs once the water and ammonia have been driven off.

The increase in stability is due to an acid functionalized protective colloid (preferably JONCRYL®-type resin) within the latex itself. The protective colloid prevents the latex particles from interacting with the rest of the ink until it reaches the oven, thus affording shelf stability. The latex containing the protective colloid is termed a supported latex. Therefore, a supported latex gives greater stability than an unsupported latex in a low VOC web offset heatset ink.

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## Example 1 - Preparation and Testing of a Comparative Web Offset Heatset Ink

A web Offset heatset ink was prepared using the ingredients listed below which include Lucidene 612, an acrylic/styrene latex that does not contain amine functionality.

20	Ingredients:	_%
	1. Raven 760 (Black pigment, Columbian	10.5
	Chemicals Company)	
-	2. RP-305 (phenolic modified rosin ester,	13.3
	acid #132, softening point - 128°C, Westvaco)	
25	3. VSPR-75 (phenolic modified rosin ester,	3.2
	acid #18, softening point 13.2 - 146-166°C,	
	Ascona/Akzo Nobel)	
	4. Lucidene 612 (acrylic/styrene emulsion,	28.3
	Rohm & Haas)	28.3
30	5. Ethylhexyltallate (Chemol)	<u>34.7</u>
		100.0

Briefly, the RP-305, VSPR-75, and ethylhexyltallate were combined in an ointment tin and heated on a hot plate with stirring until molten and homogeneous. The mixture was allowed to cool to room temperature. Then, the Raven 760 was dispersed in the mixture and grinded on a 3-roll mill until it became smooth and well dispersed. The mixture was then placed in a glass iar where the Lucidene 612 was added with stirring and mixed thoroughly.

The tack of the ink was measured on the Go Technology Digital Inkometer at a speed of 1200 rpm. The tack after one minute was 12.1 gram- meters. The viscosity of the ink was measured on the automated Duke Viscometer. The viscosity of the ink was 210 poise. The ink was printed on Rochester Institute of Technology's (R.I.T.s) Harris M -1000 web offset heatset press at speeds up to 1200 feet per minute (fpm), web exit temperature of 300°F, and with

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5 Rosos KSP 500 M-3 rountain solution. The pH of the fountain solution was 3.7 and the conductivity was 1950 mhos. This ink dried at speeds up to 500 fpm at a web exit temperature of 300°F.

Example 2 - Preparation of an Experimental Acrylic/Styrene Latex Polymer

An experimental acrylic/styrene latex polymer containing amine functionality and protective colloid was prepared from the following list of ingredients.

		<u>INGREDIENTS</u>	
	A) Deionized water	457.1g	
	B) JONCRYL resin 679*	116.5g	
15	C) NH <sub>3</sub> (30%, aqueous)	23.4g	
	D) RHODOPON UB **	3.0g	
	E) POLY G-D1200***	9.9g	
	F) Ammonium persulfate	3.9g	
	G) Deionized water	39.5g	
20	H) Dimethylaminoethyl methacrylate 74.	0g	
	I) Styrene	262.0g	
	J) 2-Ethylhexyl acrylate	0.5g	
	K) Ammonium persulfate	1.0g	
	L) Deionized water	10.0g	
2.5			

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The polymerization was carried out under a nitrogen blanket in a 4-neck 2.0-liter round bottom flask at a temperature of 88°C.

Briefly, ingredients A, C, D, and E were charged and heated to 88°C. When the temperature reached between 75 and 80°C, ingredient B was added over a 20 to 30 minute period. After all of B is added, the temperature was held for 50-60 minutes at 88°C in order for all of B to dissolve. Then, 50% of F & G ingredient mixture was added followed by 10% of H, I, & J ingredient mixture.

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<sup>\*</sup>JONCRYL 679 is a styrene acrylic resin available from Johnson Polymer that is water soluble in the presence of ammonia.

<sup>\*\*</sup>RHODOPON UB: sodium lauryl sulfate, 30% aqueous (Rhodia, Inc.).

<sup>\*\*\*</sup>POLY G-D1200: Polypropylene glycol, molecular weight 1200 (BASF).

Twenty minutes thereafter, the remaining F & G ingredient mixture and H, I, & J ingredi ent mixture were added over 1.5 hours. An hour later, a K & L ingredient mixture was added over 20 minutes and the temperature was held at 88°C for 1 hour then cooled to room temperature to discharge.

The resulting latex emulsion had a pH of 8.28 with solids averaging 46.6% (2 readings. The Brookfield viscosity was measured at 154 cP (spindle 3, 60 rpm, 25°C).

# Example 3- Preparation and testing of Experimental Black Ink

An experimental Black Ink was prepared from the following list of ingredients.

15	Ingredient	<u>%</u>
	<ol> <li>VSPR-75 (phenolic modified rosin ester,</li> </ol>	11.5
	acid #18, softening point-146-166°C,	
	Ascona/Akzo Nobel)	
	2. RP-305 (phenolic modified rosin ester,	11.5
20	acid #132, softening point-128°C, Westvaco)	
	3. Ethylhexyltallate (Chemol)	31.6
	4. Raven 760 (Black pigment, Columbian	12.0
	Chemicals Company)	
	5. Alkali blue flush (BASF Corp.)	2.6
25	6. Microcrystalliine wax compound	1.4
	(Carroll Scientific)	
	7. PTFE compound (Shamrock Technologies)	0.4
	8. Experimental Latex of Example 2	29.0
	•	100.0

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Briefly, the RP-305, VSPR-75 and ethylhexyltallate were combined in an ointment tin and heated on a hot plate with stirring until the mixture becomes molten and homogeneous. The mixture was then allowed to cool to room temperature. The Raven 760 was then dispersed in the mixture and grinded on a 3-roll mill until it became smooth and well dispersed. The mixture was placed in a glass jar where Add the alkali blue, microcrystalline wax compound, and PTFE compounds were added with stirring until they were mixed thoroughly. The experimental latex of Example 2 was then added to the mixture in the jar with stirring and mixed thoroughly to form the ink

The tack of the ink was measured on the Go Technology Digital Inkometer at a speed of 1200 rpm. The tack after one minute was 8.7 gram -meters. The viscosity of the ink was measured on the automated Duke Viscometer. The viscosity of the ink was 80 poise. The ink was printed on Rochester Institute of Technology's (R.I.T.s) Harris M -1000 web offset heatset

press, with a web exit temperature of 300°F, and with Anchor Premium MEXH IIS fountain solution. The pH of the fountain solution was 4.3 and the conductivity was 1900 mhos. This experimental ink dried at 1000 fpm at a web exit temperature of 300°F.

## Example 3- Preparation and testing of Experimental Cyan Ink

An experimental Cyan Ink was prepared from the following list of ingredients.

10 <u>Ingredients</u>	_%
<ol> <li>VSPR-75 (phenolic modified rosin ester,</li> </ol>	12.1
acid #18, softening point-146-166°C,	
Ascona/Akzo Nobel)	
2. RP-305 (phenolic modified rosin ester,	12.1
acid #132, softening point-128°C, Westvaco)	
3. Ethylhexyltallate (Chemol)	32.5
4. Blue Pigment B-15:3 (Sun Chemical)	12.0
5. Microcrystalline wax compound	1.7
(Carroll Scientific)	
20 6. PTFE compound (Shamrock Technologies)	0.6
7. Experimental Latex of Example 2	<u>29.0</u>
	100.0

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Briefly, the RP-305, VSPR-75 and ethylhexyltallate were combined in an ointment tin and heated on a hot plate with stirring until the mixture becomes molten and homogeneous. The mixture was allowed to cool to room temperature. The Pigment Blue 15:3 was dispersed in the mixture and grinded on a 3-roll mill until it became smooth and well dispersed. The mixture was placed in a glass jar where the microcrystalline wax compound, and PTFE compound were added to the mixture in with stirring and mixed thoroughly. The Experimental Latex of Example 2 was added to the mixture in the jar with stirring and mixed thoroughly to form the ink.

The tack of the ink was measured on the Go Technology Digital Inkometer at a speed of 1200 rpm. The tack after one minute was 9.2 gram -meters. The viscosity of the ink was measured on the automated Duke Viscometer. The viscosity of the ink was 130 poise. The ink was printed on Rochester Institute of Technology (R.I.T.'s) Harris M- 1000 web offset heatset press, with a web exit temperature of 300°F, and with Anchor Premium MEXH IIS fountain solution. The pH of the fountain solution was 4.3 and the conductivity was 1900 mhos. The ink dried at 1000 fpm at a web exit temperature of 300°F.

### Example 4 - Preparation and testing of Experimental Magenta Ink

An experimental Magenta Ink was prepared from the following list of ingredients.

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5	<u>Ingredients</u>	<u>%</u>
	1. VSPR-75 (phenolic modified rosin ester,	12.4
	acid #18, softening point -146-166°C,	
	Ascona/Akzo Nobel)	
	2. RP-305 (phenolic modified rosin ester,	12.4
10	acid #132, softening point-128°C, Westvaco)	
	3. Ethylhexyltallate (Chemol)	33.7
	4. Red Pigment R-57-1 (Sun Chemical)	10.2
	•	
	5. Microcrystalline wax compound	1.7
15	(Carroll Scientific)	
	6. PTFE compound (Shamrock Technologies)	0.6
	7. Experimental Latex of Example 2	<u>29.0</u>
		100.0

Briefly, the RP-305, VSPR-75 and ethylhexyltallate were combined in an ointment tin and heated on a hot plate with stirring until molten and homogeneous. The mixture was allowed to cool to room temperature. The Red Pigment R57-1 was dispersed in the mixture and grinded on a 3-roll mill until it became smooth and well dispersed. The mixture is then placed in a glass jar where the microcrystalline wax compound, and PTFE compounds were added to it with stirring and mixed thoroughly. The experimental latex of Example 2 was added to the mixture in the jar with stirring and mixed thoroughly to form the ink.

The tack of the ink was measured on the Go Technology Digital Inkometer at a speed of 1200 rpm. The tack after one minute was 10.1 gram- meters. The viscosity of the ink was measured on the automated Duke Viscometer. The viscosity of the ink was 175 poise. The ink was printed on Rochester Institute of Technology's (R.I.T.'s) Harris M -1000 web offset heatset press, with a web exit temperature of 300°F, and with Anchor Premium MXH IIS fountain solution. The pH of the fountain solution was 4.3 and the conductivity was 1900 mhos. The ink dried at speeds up to 1600 fpm at a web exit temperature of 300°F.

## 35 Example 5 - Preparation and testing of Experimental Yellow Ink

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An experimental Yellow lnk was prepared from the following list of ingredients.

	<u>Ingredients</u>	_%
	<ol> <li>VSPR-75 (phenolic modified rosin ester,</li> </ol>	12.2
40	acid #18, softening point-146-166°C,	
	Asconal/Akzo Nobel)	
	2. RP-305 (phenolic modified rosin ester,	12.2

	VV O 2003/000252	1 € 17 € 52 00 47 0 -	10070
5	acid #132, softening point-128°C, Westvaco)		
	3. Ethylhexyltallate (Chemol)	33.5	
	4. Yellow Pigment Y-12 (Sun Chemical)	10.8	
	5. Microcrystalliine wax compound	1.7	
	(Carroll Scientific)		
10	7. PTFE compound (Shamrock Technologies)	0.6	
	8. Latex R3118-10 (acrylic/styrene emulsion,	29.0	
	Sun Chemical)	100.0	

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Briefly, the RP-305, VSPR-75 and ethylhexyltallate were combined in an ointment tin and heated on a hot plate with stirring until the mixture becomes molten and homogeneous. The mixture was then allowed to cool to room temperature. The Yellow Pigment 12 was dispersed in the mixture and grinded on a 3-roll mill until it became smooth and well dispersed. The mixture was placed in a glass jar where the microcrystalline wax compound, and PTFE compounds were added to the mixture in the jar with stirring and Mixed thoroughly. The experimental latex of Example 2 was added to the mixture in the jar with stirring and mixed thoroughly to form the ink

The tack of the ink was measured on the Go Technology Digital Inkometer at a speed of 1200 rpm. The tack after one minute was 9.7 gram -meters. The viscosity of the ink was measured on the automated Duke Viscometer. The viscosity of the ink was 150 poise. The ink was printed on Rochester Institute of Technology's (R.I.T.'s) Harris M -1000 web offset heatset press, with a web exit temperature of 300°F, and with Anchor Premium MEXH IIS fountain solution. The pH of the fountain solution was 4.3 and the conductivity was 1900 mhos. The ink dried at speeds up to 1000 fpm at a web exit temperature of 300°F.

The invention has been described in terms of preferred embodiments thereof, but is more broadly applicable as will be understood by those skilled in the art. The scope of the invention is only limited by the following claims.